



COURSE DESCRIPTION CARD - SYLLABUS

Course name

Project in inorganic chemical technology [S1TCh2>PzTCN]

Course

Field of study

Chemical Technology

Year/Semester

3/5

Area of study (specialization)

–

Profile of study

general academic

Level of study

first-cycle

Course offered in

Polish

Form of study

full-time

Requirements

elective

Number of hours

Lecture

0

Laboratory classes

0

Other (e.g. online)

0

Tutorials

0

Projects/seminars

15

Number of credit points

1,00

Coordinators

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Lecturers

Prerequisites

Student has knowledge of general and inorganic chemistry, physical chemistry and apparatus of chemical industry, knows the basic methods, techniques and tools used in chemical analysis (core curriculum of I and II year of the studies). Student can obtain information from literature, databases and other sources, can interpret the obtained information to draw conclusions and formulate opinions in the area of general and inorganic chemistry. Student is able to apply that knowledge in practice, both during the implementation work and the further education. Student is able to interact and work in a group. Student is able to properly identify the priorities used to perform a specific task. Student understands the need for further education.

Course objective

Acquiring basic knowledge in the field of inorganic chemical technology. Understanding the basic industrial processes and operations related to inorganic technology. Ability to select raw materials and chemical intermediates. Understanding the methods of obtaining inorganic products and their identification. Indication of the possibility of using products manufactured in inorganic technology processes. Proper waste handling. Proposal of using environmentally friendly technologies. Material and energy balances of selected inorganic technologies.

Course-related learning outcomes

Knowledge:

- K_W03 - has the necessary knowledge of chemistry to enable understanding of chemical phenomena and processes
- K_W07 - knows the rules of environmental protection related to inorganic chemical technology and waste management
- K_W08 - has a systematically, theoretically founded general knowledge in the field of general and inorganic chemistry
- K_W09 - has the necessary knowledge about both natural and synthetic raw materials, products and processes used in inorganic chemical technology, as well as about the directions of development of the chemical industry in the country and in the world
- K_W10 - knows the basics of thermodynamics, kinetics, surface phenomena and catalysis of chemical processes
- K_W13 - has knowledge of inorganic chemical technology and the apparatus of the chemical industry
- K_W14 - has a basic knowledge of the life cycle of products, equipment and installations in the chemical industry

Skills:

- K_U01 - can obtain the necessary information from literature, databases and other sources related to chemical sciences, correctly interprets them, draws conclusions, formulates and justifies opinions
- K_U02 - can work both individually and as a team in a professional and other environment
- K_U04 - can prepare and present in Polish an oral presentation on chemical technology
- K_U05 - has the ability to self-study
- K_U16 - based on general knowledge, explains the basic phenomena associated with significant processes in inorganic chemical technology
- K_U18 - distinguishes between types of chemical reactions and has the ability to select them for chemical processes
- K_U22 - knows the physical and chemical properties of chemical compounds and materials
- K_U25 - assesses the risks associated with the use of chemical products and processes

Social competences:

- K_K01 - understands the need for further training and raising their professional, personal and social competences
- K_K02 - is aware of the importance and understanding of non-technical aspects and effects of engineering activities, including their impact on the environment and the associated responsibility for decisions made
- K_K03 - is able to cooperate and work in a group, inspire and integrate engineering environments

Methods for verifying learning outcomes and assessment criteria

Learning outcomes presented above are verified as follows:

Stationary: presentation of material on a topic given by the teacher; solving scientific problems, assessment of student's activity on projects, assessment of implementation and solving of given project tasks, assessment of team work and self-presentation skills. The final assessment will include the ability to present a given scientific issue and a discussion with the teacher and other students on a given topic.

Online: presentation of material on a topic given by the teacher using the eMeeting or Zoom platform; solving scientific problems, assessment of student's activity on projects, assessment of implementation and solving of given project tasks, assessment of team work and self-presentation skills. The final assessment will include the ability to present a given scientific issue and a discussion with the teacher and other students on a given topic.

Programme content

1. Mineral and fuel resources.
2. The method of obtaining energy for technological processes.
3. Technology of sulfur compounds.
4. Technology of phosphorus compounds.
5. Technology of nitrogen compounds.
6. Production of soda.
7. Wet and dry methods of enrichment of minerals.
8. Introduction into advanced inorganic technology.

Course topics

none

Teaching methods

Projects - multimedia presentations on the topic given by the teacher, solving research problems.

Bibliography

Basic:

1. K. Schmidt-Szałowski, J. Sentek, J. Raabe, E. Bobryk, Podstawy technologii chemicznej. Procesy w przemyśle nieorganicznym, Oficyna Wydawnicza Politechniki Warszawskiej Warszawa 2004.
2. J.A. Moulijn, M. Makkee, A. van Diepen: Chemical Process Technology, Wiley-Blackwell, Chichester 2013.
3. J. Szarawara, J. Piotrowski, Podstawy teoretyczne technologii chemicznej, WNT Warszawa 2010.

Additional:

1. C.H. Bartholomew and R.J. Farrauto, Fundamentals of industrial catalytic processes, Wiley, Hoboken, New Jersey 2006.
2. M.B. Hocking, Handbook of chemical technology and pollution control, Elsevier, Amsterdam 2005.
3. G. Ertl, H. Knözinger, F. Schüth, J. Weitkamp, Handbook of heterogeneous catalysis, WILEY-VCH Weinheim 2008.
4. S. Bretsznajder, W. Kawecki, J. Leyko, R. Marcinkowski: Podstawy ogólne technologii chemicznej, WNT, Warszawa 1973.
5. M. Taniewski: Technologia chemiczna - surowce, Wydawnictwo Politechniki Śląskiej, Gliwice 1997.
6. H. Konieczny: Podstawy technologii chemicznej, PWN, Warszawa 1975.
7. J. Kępiński: Technologia chemiczna nieorganiczna, PWN, Warszawa 1975.
4. Laboratory materials
8. Laboratory materials

Breakdown of average student's workload

	Hours	ECTS
Total workload	25	1,00
Classes requiring direct contact with the teacher	15	0,50
Student's own work (literature studies, preparation for laboratory classes/ tutorials, preparation for tests/exam, project preparation)	10	0,50